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It has been mentioned as an elementary reaction in the problem statement but in the proposed solution the rate law is based on the reaction equation that has been divided by stoichiometric coefficient of A. (2) The value of calculated is incorrect. $= yA_0^0 = 0.6 (3+5-2-3) = 1.8$ is the correct value.

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Solutions Manual; Review about the book: Octave Levenspiel was a professor of the field Chemical engineering at Oregon State University. In this vast and evergreen field his major interests lied in Chemical Reaction Engineering which is one of the core subjects in Chemical Engineering.

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